

# Worldwide Pollution Control Association

Gulf Power Coal to Gas Seminar  
May 30-31, 2012

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# SCR/CO Catalyst Considerations



Johnson Matthey  
Catalysts



**WPCA/Gulf Power  
Coal to Gas Conversion Seminar  
Pensacola, FL  
May 30-31, 2012**

Ken Jeffers

ENVIRONMENTAL CATALYSTS AND TECHNOLOGIES



# Topics



- Regulatory Drivers for Controlling Emissions – NO<sub>x</sub>, CO
- SCR Catalyst for NO<sub>x</sub> control
- CO oxidation catalyst



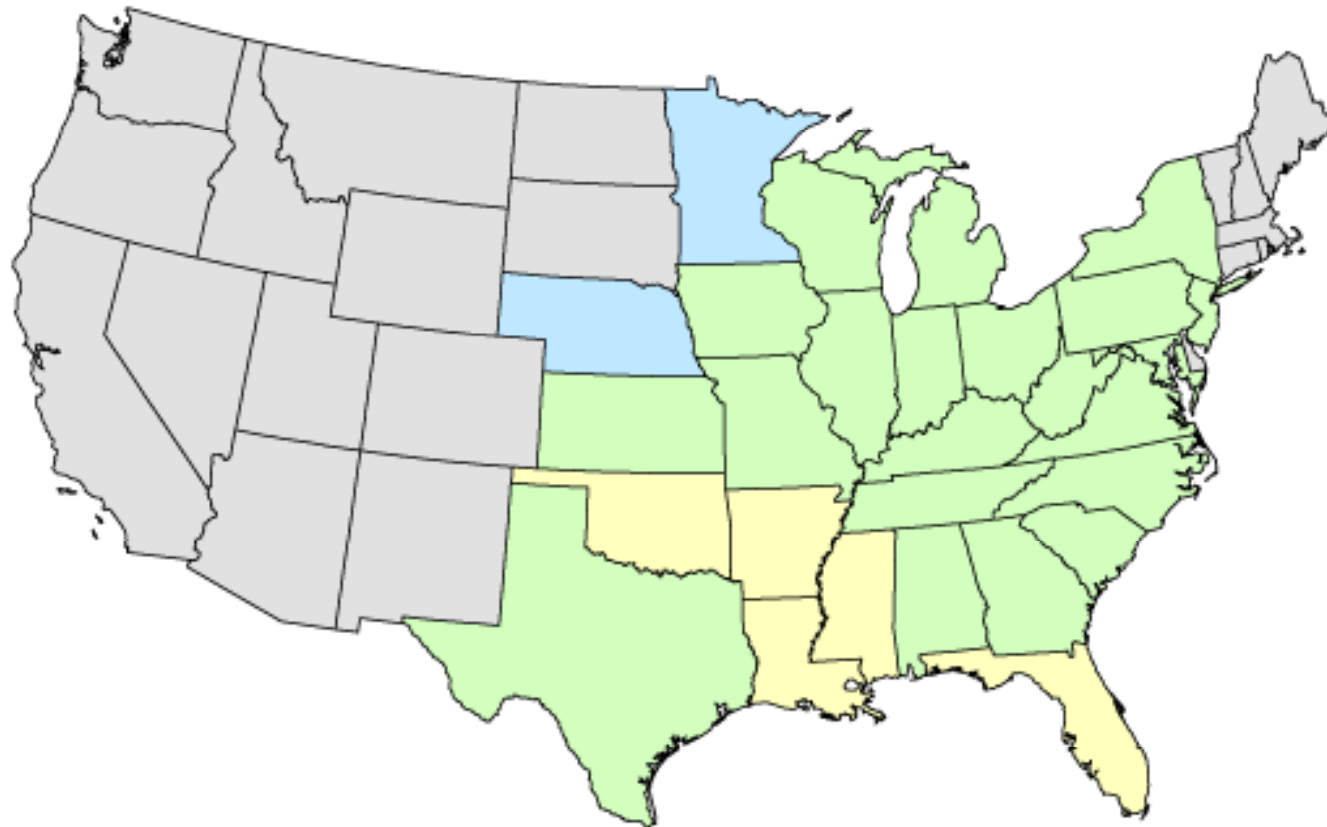
# Regulatory Drivers - NO<sub>x</sub> Control



- NO<sub>2</sub> is 1 of 6 criteria pollutants
- Respiratory irritant, contributes to low-level ozone formation
- Regulated per EPA NAAQS
- MATS, Amendments to NSPS: NO<sub>x</sub> emission limit – 0.7 lb/MWh
- Further controls necessary per CSAPR which requires:
  - 23 states to reduce SO<sub>2</sub> and NO<sub>x</sub> to help downwind states attain 2006 24-hour and/or 1997 annual PM<sub>2.5</sub> NAAQS
  - 20 states to reduce Ozone Season NO<sub>x</sub> to help downwind areas attain 1997 8-hour Ozone NAAQS



# CSAPR States



- States controlled for both fine particles (annual SO<sub>2</sub> and NO<sub>x</sub>) and ozone (ozone season NO<sub>x</sub>) (21 States)
- States controlled for fine particles only (annual SO<sub>2</sub> and NO<sub>x</sub>) (2 States)
- States controlled for ozone only (ozone season NO<sub>x</sub>) (5 States)
- States not covered by the Cross-State Air Pollution Rule



# Regulatory Drivers - CO Control



- CO is 1 of 6 criteria pollutants
- Majority of CO emissions come from mobile sources
- Contributes to low-level Ozone formation  
 $\text{CO} + 2\text{O}_2 \rightarrow \text{CO}_2 + \text{O}_3$
- NAAQS for CO  
8-hour average:  $\leq 9$  ppmv  
1-hour average:  $\leq 35$  ppmv



# Fossil Fuel Air Emissions Comparison (lb/MMBTU of Energy Input)



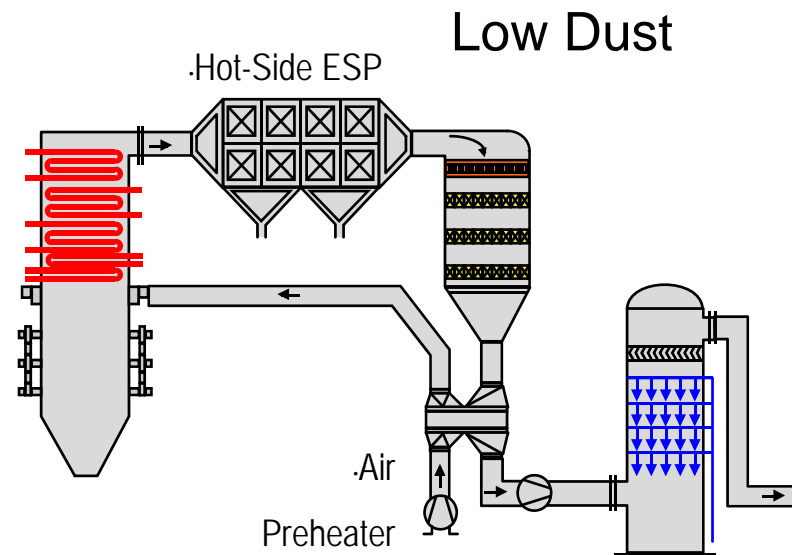
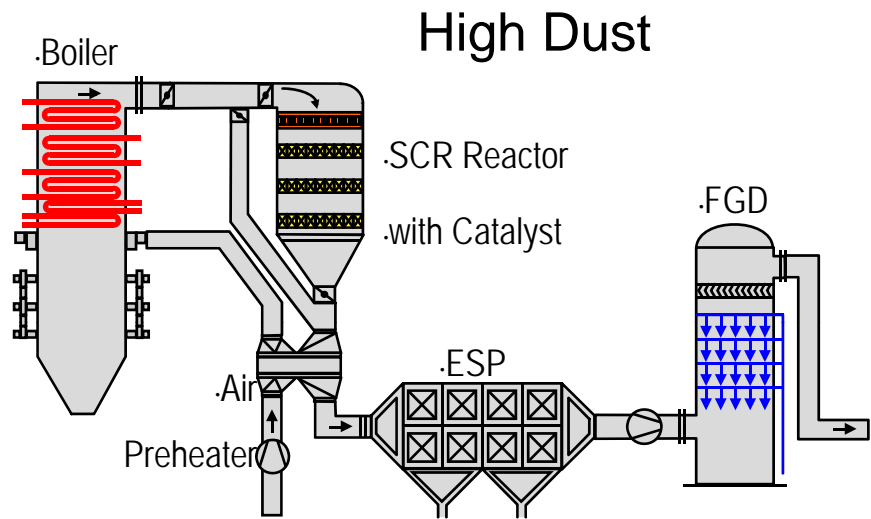
Pollutant	Natural Gas	Oil	Coal
Carbon Dioxide	117	164	208
Carbon Monoxide	0.040	0.033	0.208
Nitrogen Oxides	0.092	0.448	0.457
Sulfur Dioxide	0.001	1.122	2.591
Particulates	0.007	0.084	2.744
Mercury	0.000	$0.007 \times 10^{-3}$	$0.016 \times 10^{-3}$



# SCR Catalyst for NO<sub>x</sub> Control



# SCR Configuration in Power Plants



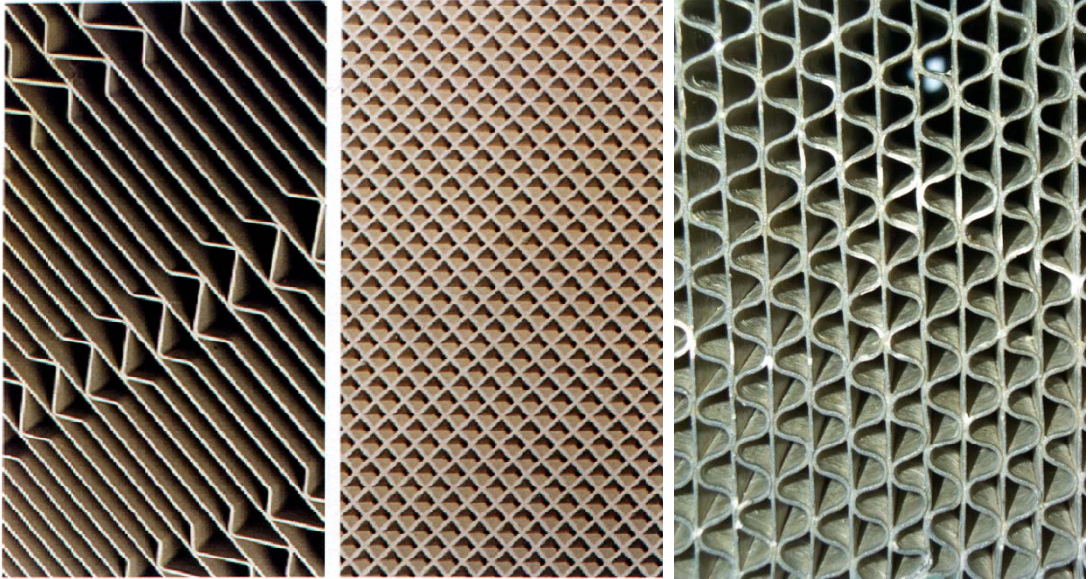
# SCR Catalyst Types



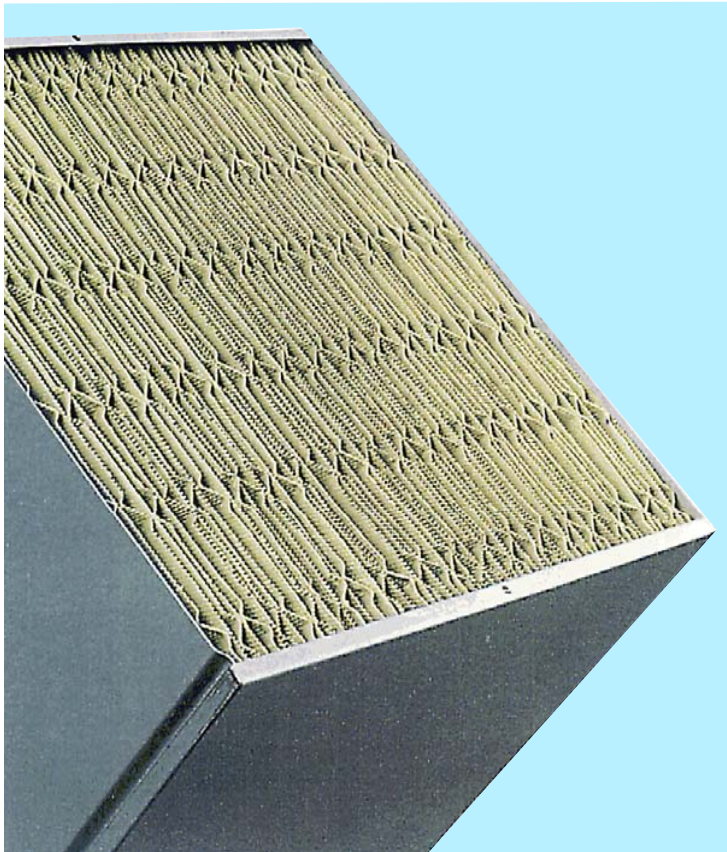
Plate

Honeycomb

Corrugated



# Plate-type Catalyst



## Composition

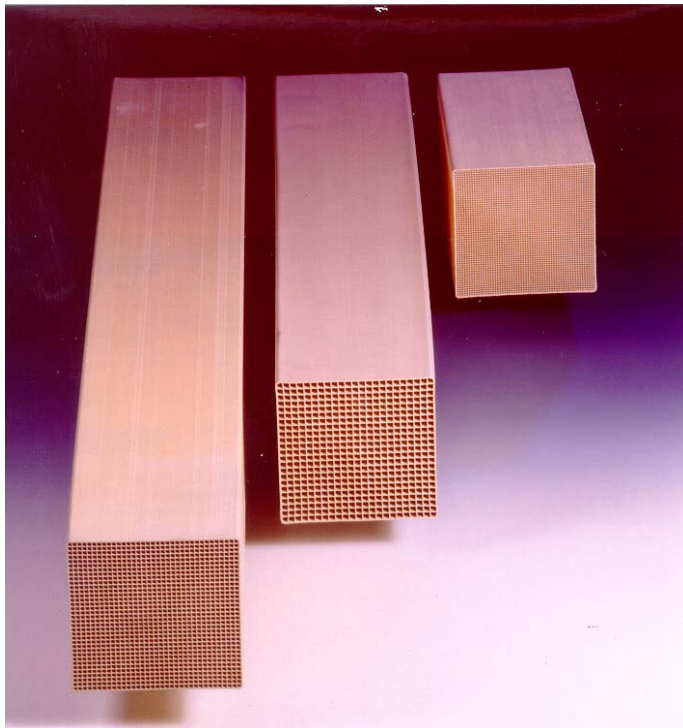
- Stainless steel carrier, ceramic material rolled on
- $\text{TiO}_2$ , V-oxide/W-oxide/Mo-oxide
- Notches formed into plates provide separation
- Inserted in element boxes with variable spacing: 60 to 90 plates
- Variable plate height up to 700 mm

## Advantages

- Ideal for high dust configurations
- Plugging, erosion resistance
- Low pressure loss



# Honeycomb Catalyst



## Composition

- Homogeneously extruded ceramic with square-opening cell structure
- $\text{TiO}_2$ , V-oxide/W-oxide
- Variable block height up to 1300 mm
- Pitch: 2.1 mm – 9.2 mm

## Advantages

- Ideal for low/no-dust applications
- High active surface area per unit volume



# Catalyst Pitch and Structure

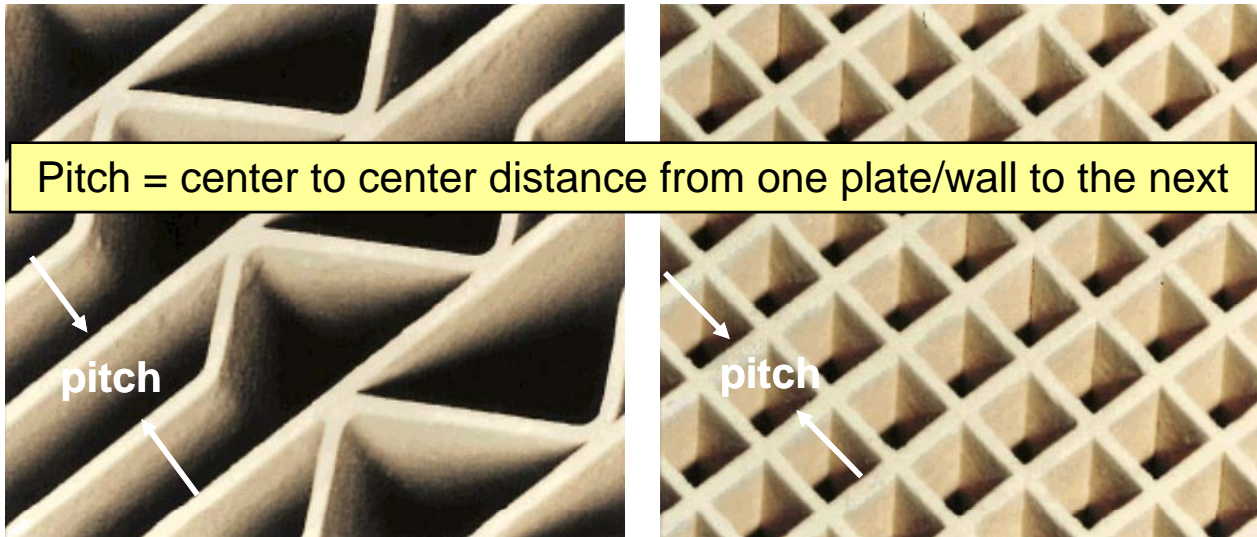


Plate-Type Structure

- Flexible plates
- Rectangular openings
- Pitch: 5 to 7 mm

Honeycomb Structure

- Rigid structure
- Square openings
- Pitch: 2.1 to 9.2 mm



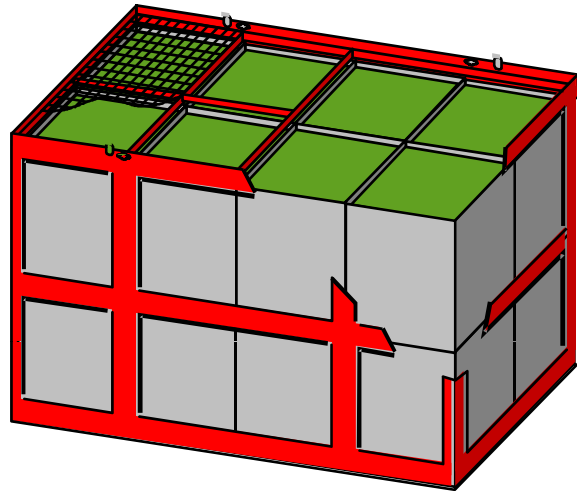
# Pitch Selection by Application



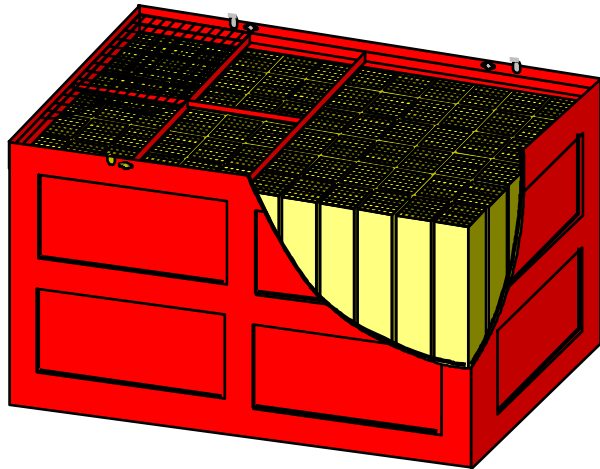
Application	Plate Pitch	Honeycomb Pitch
Natural Gas, Diesel	-	2.1 mm (70 x 70 cell) – 3.7 mm (40 x 40 cell)
Tail-End, Low-dust	5.0 – 5.6 mm	3.7 mm (40 x 40 cell) – 6.7 mm (22 x 22 cell)
Fuel Oil	5.0 – 5.6 mm	4.9 mm (30 x 30 cell) – 6.4 mm (23 x 23 cell)
Bituminous Coal	5.6 – 6.0 mm	6.7 mm (22 x 22 cell) – 8.2 mm (18 x 18 cell)
PRB Coal	5.6 – 7.0 mm	8.2 mm (18 x 18 cell) – 9.2 mm (16 x 16 cell)
Lignite	6.0 – 7.0 mm	? (no US data)



# Catalyst Modules for Utility SCRs



- Catalyst elements arranged in steel frames
  - Plate – 2 levels of 8 element boxes
  - Honeycomb – 72 monoliths
- Standardized cross-section
- Possible to interchange catalyst types within reactor
- Module height varies with catalyst height



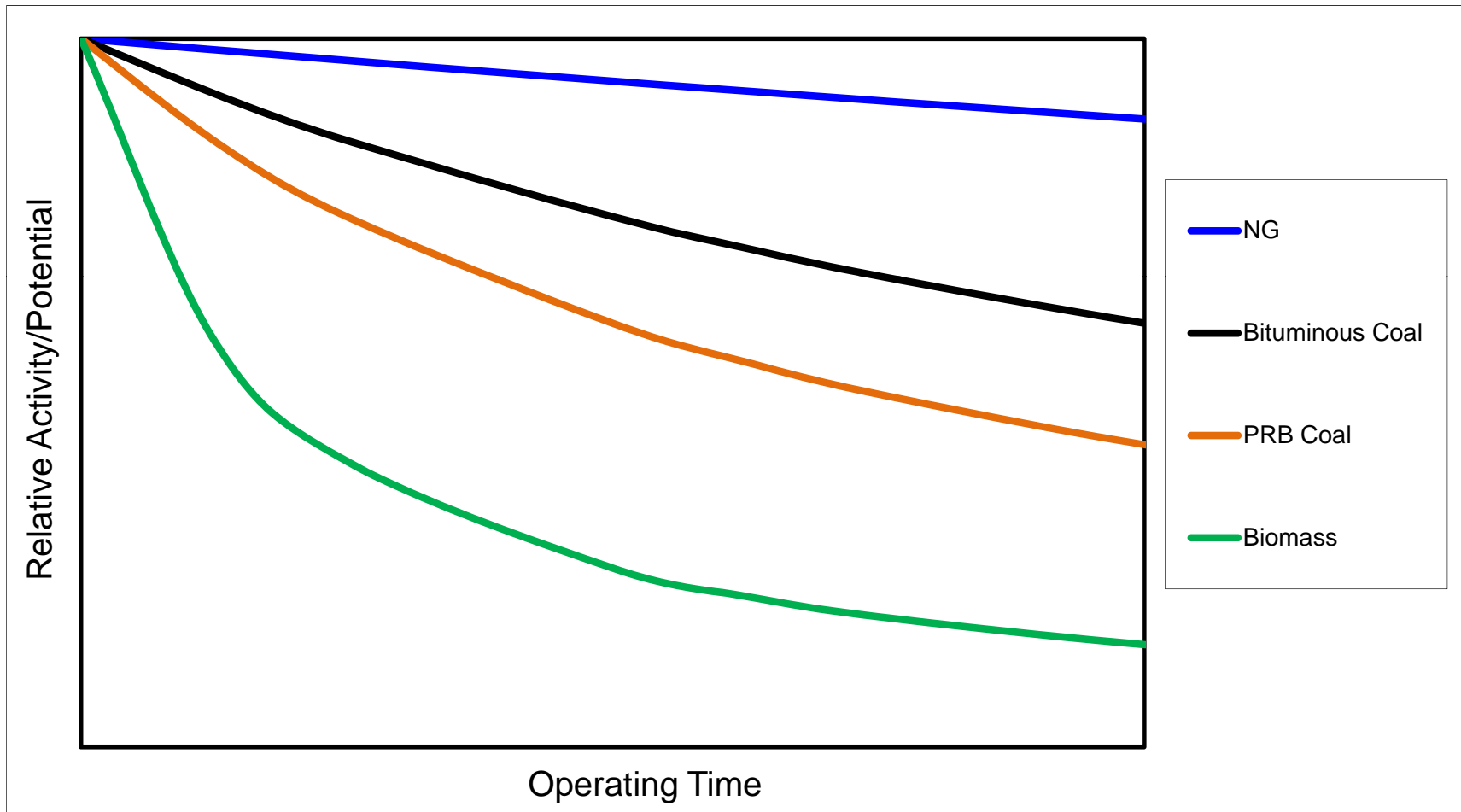
# Suitable Operating Conditions



- Flue Gas Temperature: 600 – 800 °F
- Flue gas linear flow velocity: 5.0 – 6.0 m/s
- Thorough  $\text{NH}_3$ - $\text{NO}_x$  mixing: 5% RMS (required for >85% de $\text{NO}_x$  with low  $\text{NH}_3$  slip)
- For flue gas with particulate matter, need
  - Popcorn ash/LPA screens upstream of SCR reactor
  - Soot-blowers or sonic horns at catalyst layers
- Means to keep catalyst dry and frost-free during outage periods

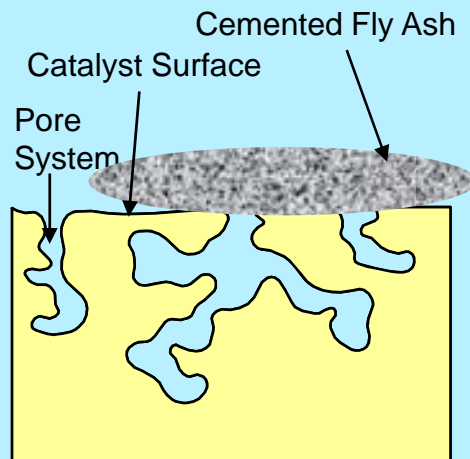


# SCR Catalyst Deactivation

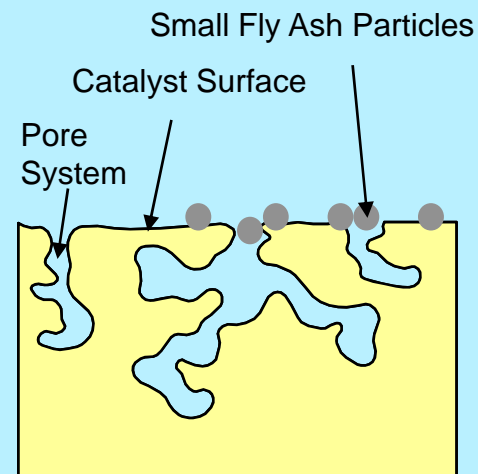


# Catalyst Deactivation Mechanisms

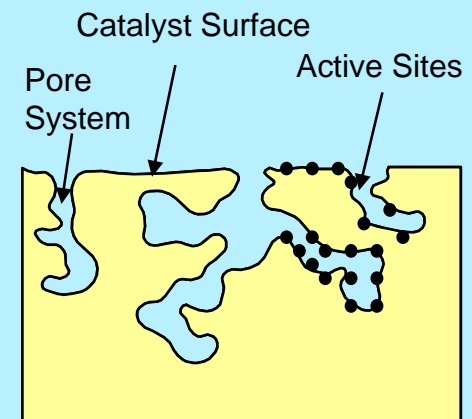
**Masking:**  
Macroscopic blockage  
of catalyst surface  
by cemented fly ash



**Plugging:**  
Microscopic blockage of  
pore system  
by small fly ash particles



**Poisoning:**  
Deactivation of active  
sites by chemical attack



# Advantages with Gas-firing



- Catalyst Selection For Coal . . .
  - Large pitch to avoid ash plugging
  - Minimized  $\text{SO}_2$ - $\text{SO}_3$  oxidation
  - Frequent replacement – every 2 years
- For Natural Gas firing . . .
  - Much slower catalyst deactivation – longer catalyst life
  - Fewer catalyst replacements
  - No fly ash plugging
  - No Sulfur → no ABS, can tolerate higher  $\text{NH}_3$  slip
  - Possible reduced  $\text{NO}_x$  reduction requirements
  - If 100% gas, can use higher cell density catalyst – reduced volume requirement



# Sizing Examples



	Base Bit Coal Case	NG – equal NO <sub>x</sub> load	NG – reduced NO <sub>x</sub> load
NO <sub>x</sub> in	300 ppmvd	300 ppmvd	240 ppmvd
NO <sub>x</sub> out	40 ppmvd	40 ppmvd	40 ppmvd
NO <sub>x</sub> Reduction	87%	87%	83%
NH <sub>3</sub> slip	2 ppmvd	2 ppmvd	5 ppmvd
Rel P0	1.00	0.83	0.77
Catalyst Pitch	5.6 mm plate	3.7 mm HC	3.7 mm HC
Relative Volume	1.00	0.26	0.21
Pressure drop	1.7 iwg	1.9 iwg	1.5 iwg



# Catalyst Options for Coal to Gas Conversion



For Units already having SCR with coal-fired catalyst

- Conversion to 100% Gas firing is OK
  - Remove portion of coal-fired catalyst for reuse in other coal units
  - Replace all coal-fired catalyst with gas-fired catalyst
- If coal will be backup fuel or co-fired, gas-specific catalyst is not suitable
  - Fly ash plugging
  - High  $\text{SO}_2 - \text{SO}_3$  oxidation

If SCR required on unit after conversion

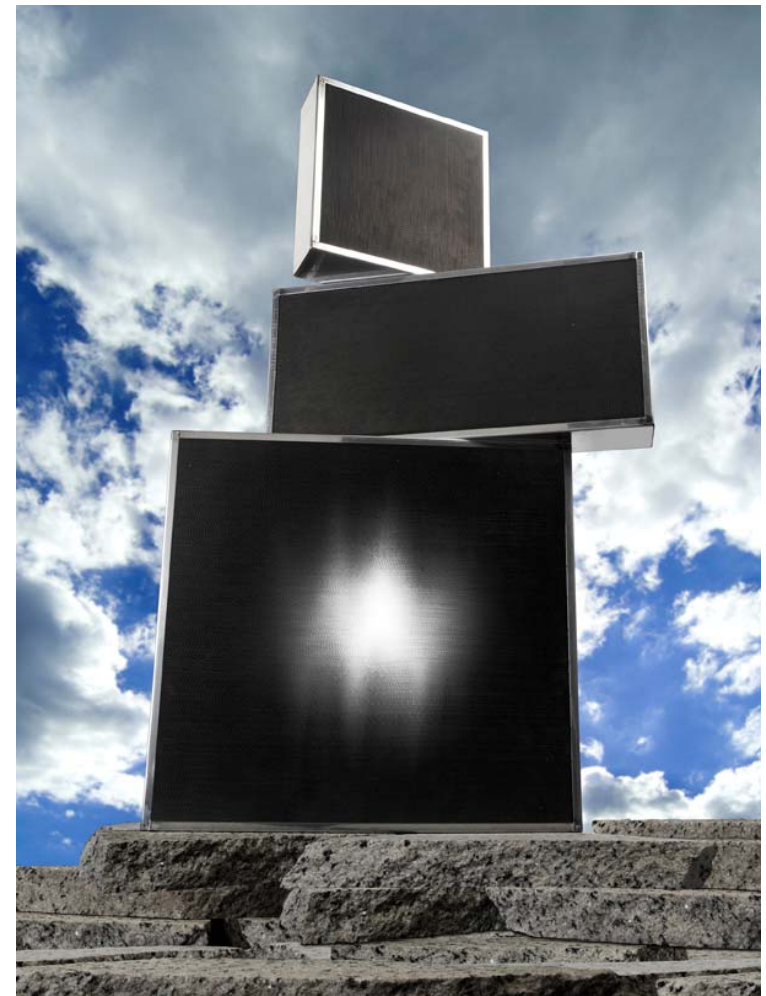
- Conversion to 100% Gas firing – install gas-fired catalyst
- Conversion to Gas with Coal backup or co-firing gas and coal – need catalyst suitable for coal-firing.



# CO Oxidation Catalyst



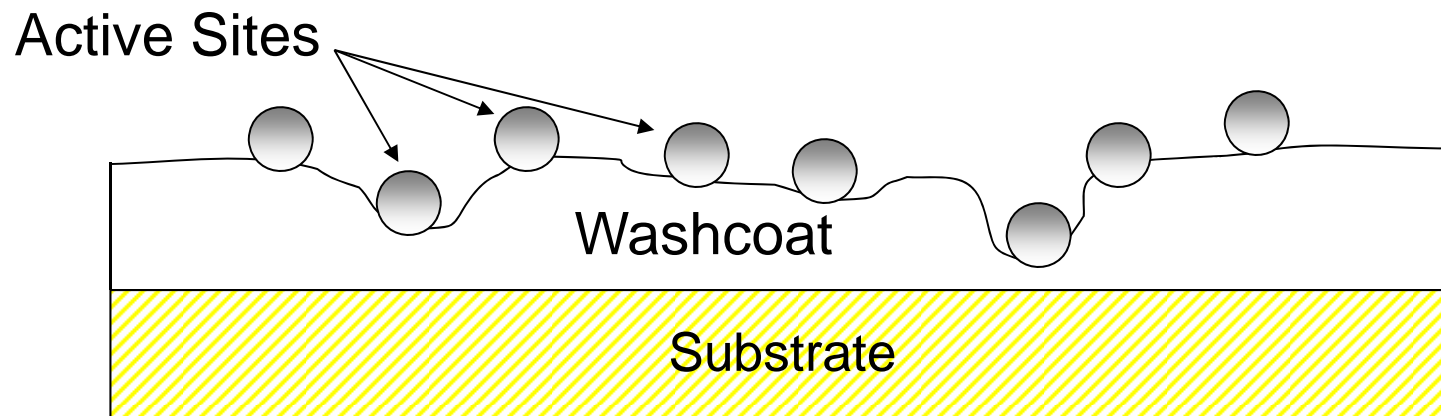
- Oxidizes CO and VOCs to CO<sub>2</sub>
- Stainless steel foil or ceramic honeycomb substrate, high surface area alumina washcoat, Platinum Group Metals (PGM) embedded in washcoat
- Blocks with 25 cpsi – 400 cpsi
- Standard block is 24" x 24", 200 cpsi metal foil for gas turbine applications
- Variable block cross-section possible
- Metal foil depths of 1.5 – 5 inches



# Catalyst Conversion Efficiency



- Catalyst “conversion efficiency” is related to the catalyst durability, reactant residence time (gas hourly space velocity) and the active surface area of the catalyst



- The number of “active sites” (active surface area) is maximized by dispersing the catalyst on substrate media coated with a high surface area washcoat

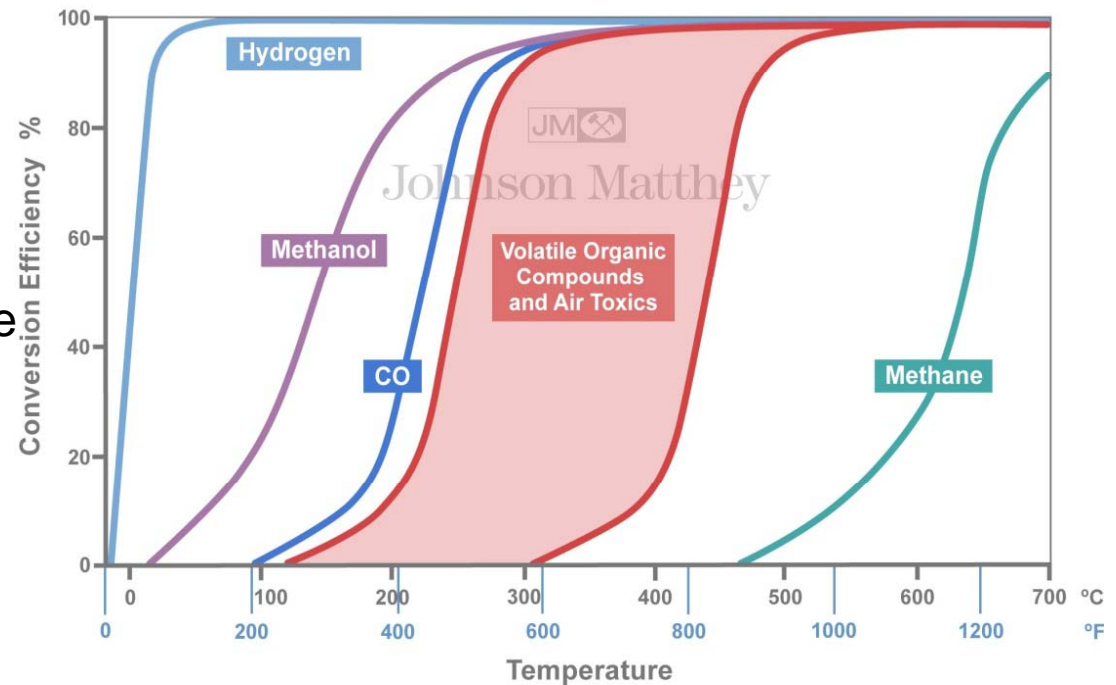




# Catalyst “Light Off”



- The activation energy required to make the reaction happen is usually supplied thermally
- Thus, all catalysts have a minimum operating temperature known as the “light-off” point
- The light-off temperature will vary according to the catalyst type and reactants present



- The temperature at which catalysts achieve 50% conversion efficiency is typically referred to as the “T50 light-off” point



# Catalyst Deactivation Mechanisms



- Thermal deactivation: *(typically irreversible)* Occurs above 1200 °F
  - Failure of substrate material
  - Sintering of active catalytic species
  - Sintering of support
  - Reaction of catalyst materials
  - Catastrophic thermal event
- Poisoning: *(typically reversible)*
  - Chemical (selective) contamination of active sites
  - Physical (non-selective) contamination
    - *masking, fouling, plugging of cells/pores, washcoat attrition*



# Catalyst Deactivators (Poisons)



- Heavy and Base Metals:
  - Lead (*Pb*)
  - Arsenic (*As*)
  - Zinc (*Zn*)
  - Copper (*Cu*)
  - Sulfur (*S*)
  - Mercury (*Hg*)
  - Tin (*Sn*)
  - Iron (*Fe*)
  - Chrome (*Cr*)
  - Phosphorous (*P*)
  - Silicon (*Si*)
  - Nickel (*Ni*)
  - Antimony (*Sb*)
- High Molecular Weight Organic Material
- Dust and Particulates
- NOT SUITABLE FOR COAL-FIRED APPLICATIONS



# Selective (Chemical) Poisoning



- Example – Sulfur can react directly with active site and degrade performance.



Alloy has low activity  
(For certain poisons it  
is possible to reactivate  
the catalyst by thermal  
treatment)



# Catalyst Regeneration or Cleaning Options



- **Thermal**

- **Elevating Temperature**

- *Oxidizes organic material*

- **Physical**

- **Vacuuming or Compressed Air Blowing**

- *Removes dust and debris*

- **Chemical (including DI Water)**

- **Washing**

- *Removes masking agents*
    - *Restores surface area*
    - *Reduces poison concentration*



# CO Catalyst Options if required after Coal to Gas Conversion



- Suitable for units converted to 100% Gas firing
- Not suitable for coal firing (or fuels containing particulate)
  - Back-up fuel or co-firing
- Suitable for low S fuel oil / ULSD / Light oil used as back-up fuel.
  - Firing should be limited – Typically < 500 hours per year
- Not suitable for HFO



# Summary Points



- Gas-firing has many advantages for SCR
  - Slower deactivation – longer lifetime
  - Low/No Sulfur → No ABS, higher  $\text{NH}_3$  slip tolerable
  - No fly ash/particulate plugging
  - May result in Lower  $\text{NO}_x$  load, lower de $\text{NO}_x$  requirement
  - Low pitch, high cell density catalyst elements – lower volume requirement as compared to coal applications
- Catalyst sized for natural gas not suitable for coal-firing
- CO Oxidation catalyst suitable for gas-firing and low/no sulfur fuels
- High rates of CO oxidation are possible
- CO Oxidation catalyst is not suitable for coal-firing or fuels with ash/particulate



# Thank You!



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